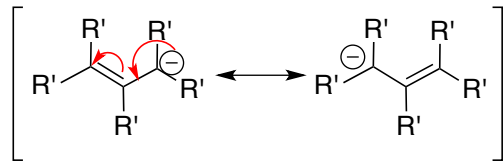
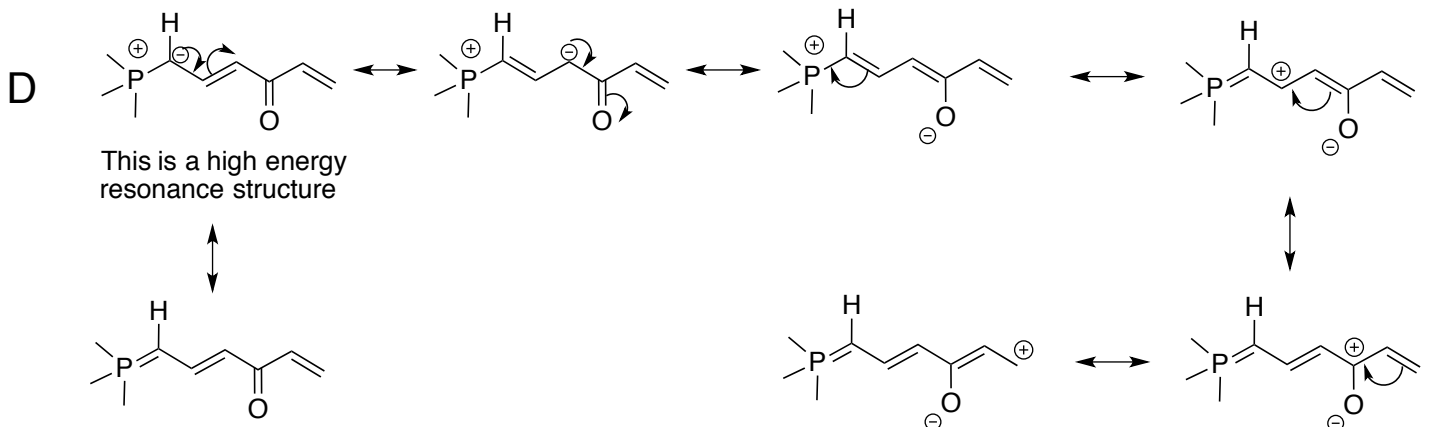
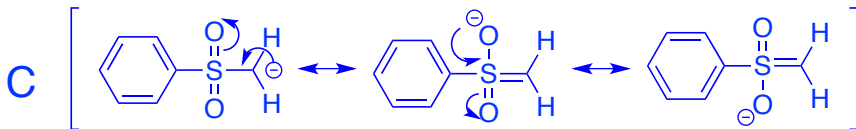
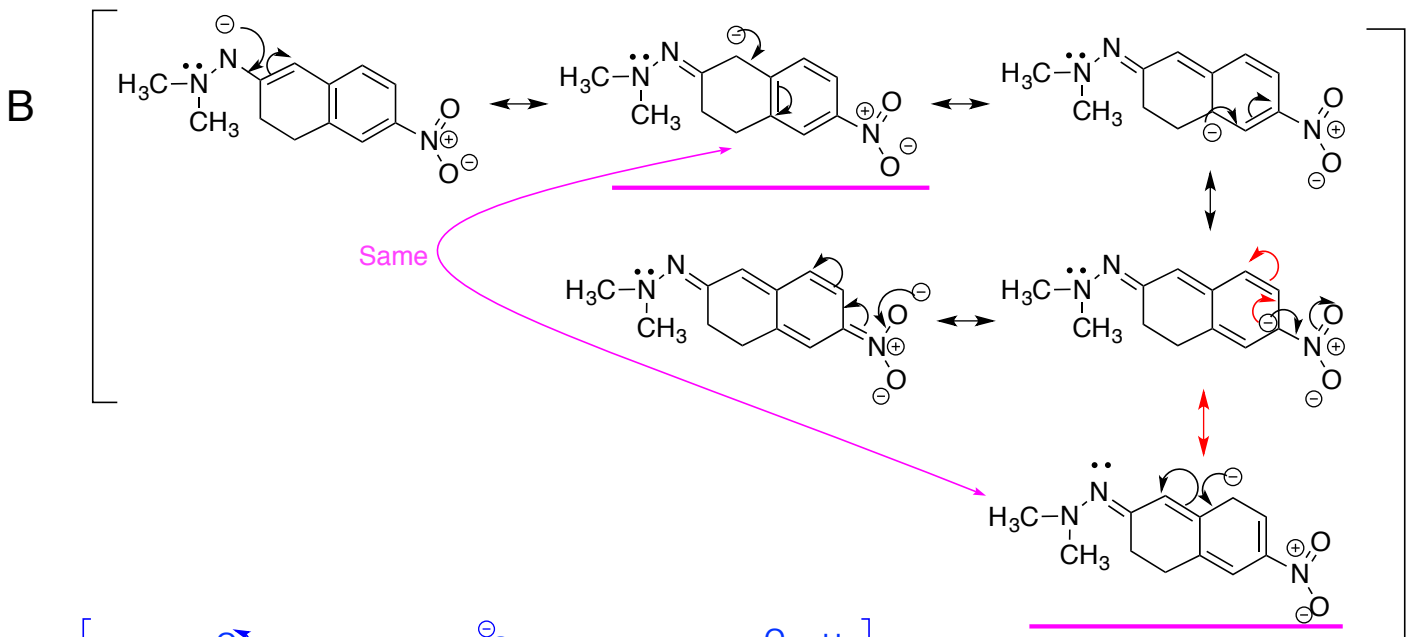
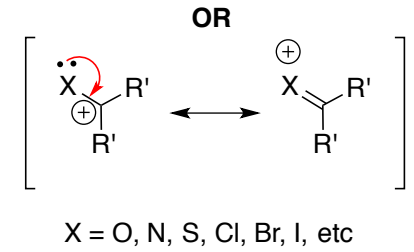
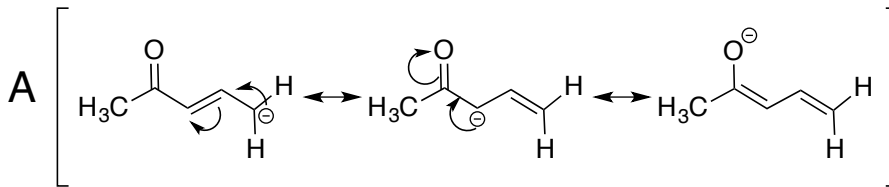
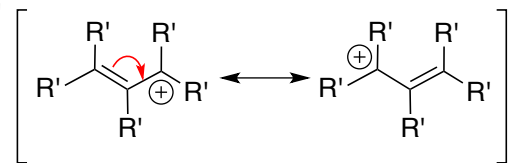
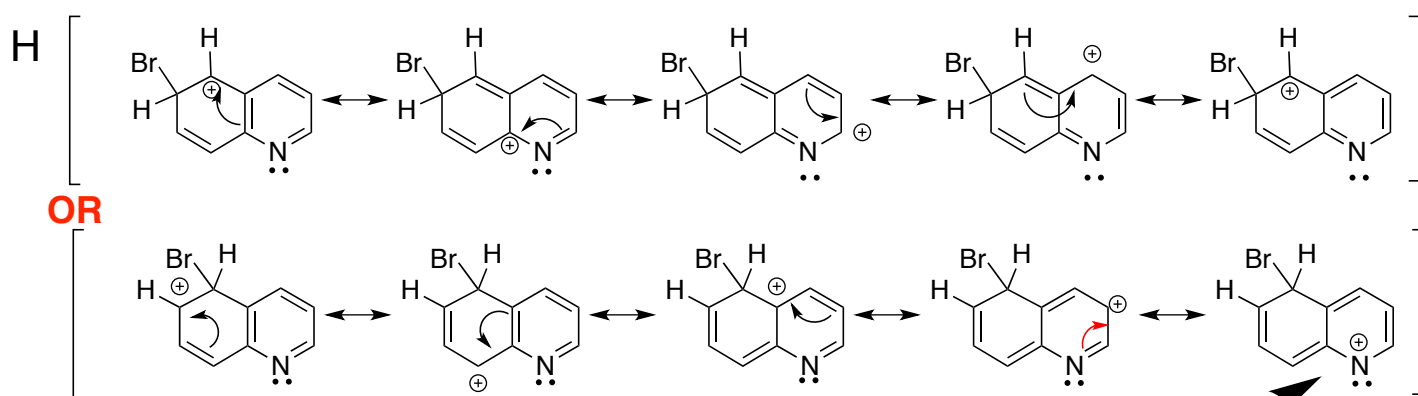
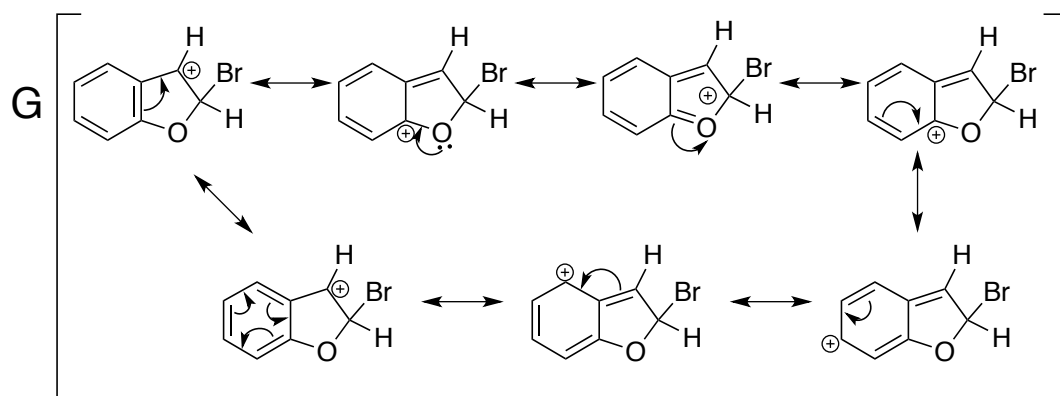
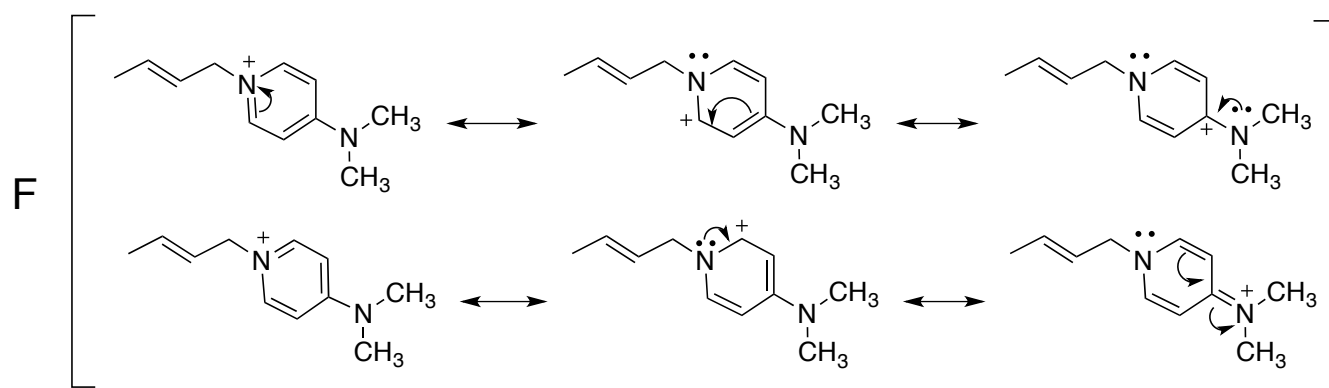
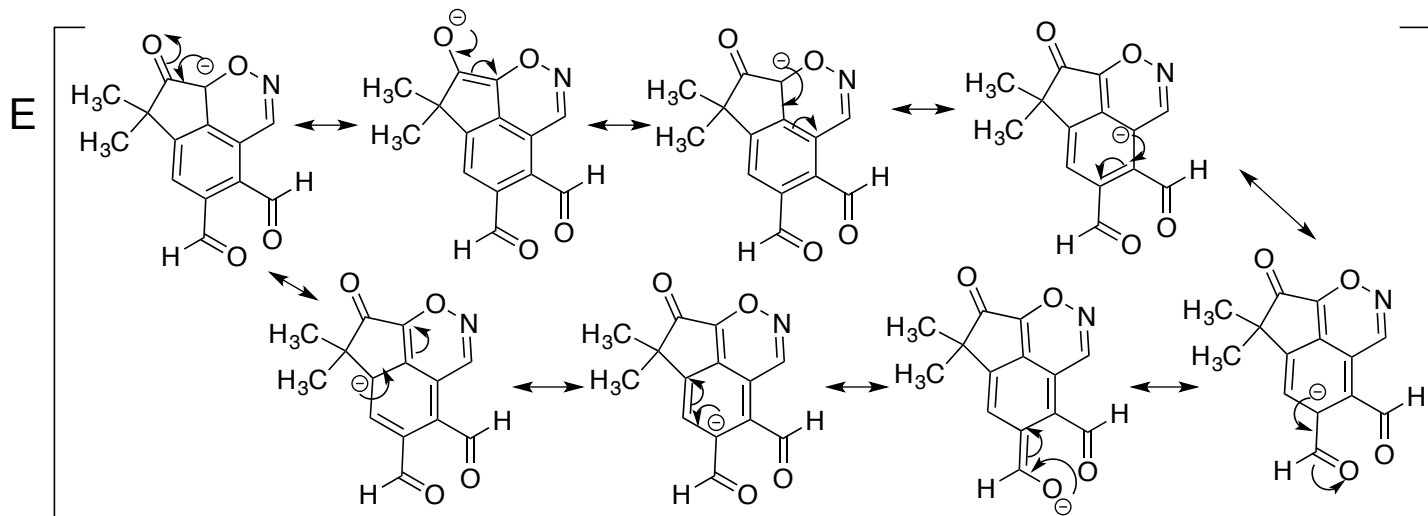


Anion Resonance Structures

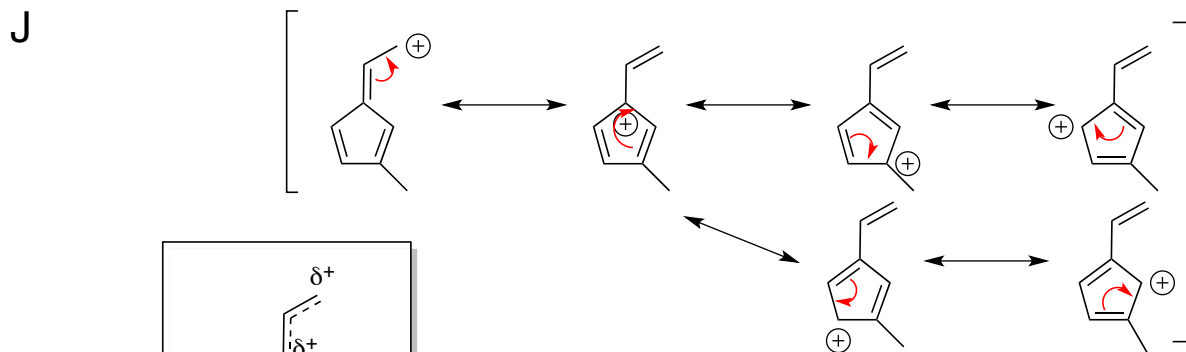
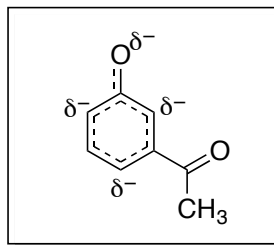
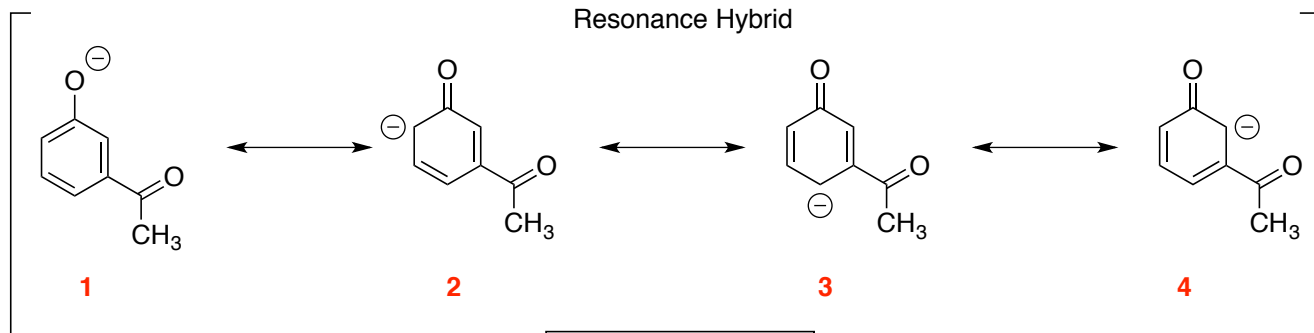
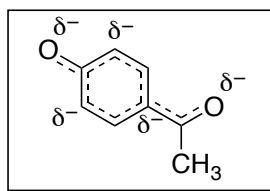
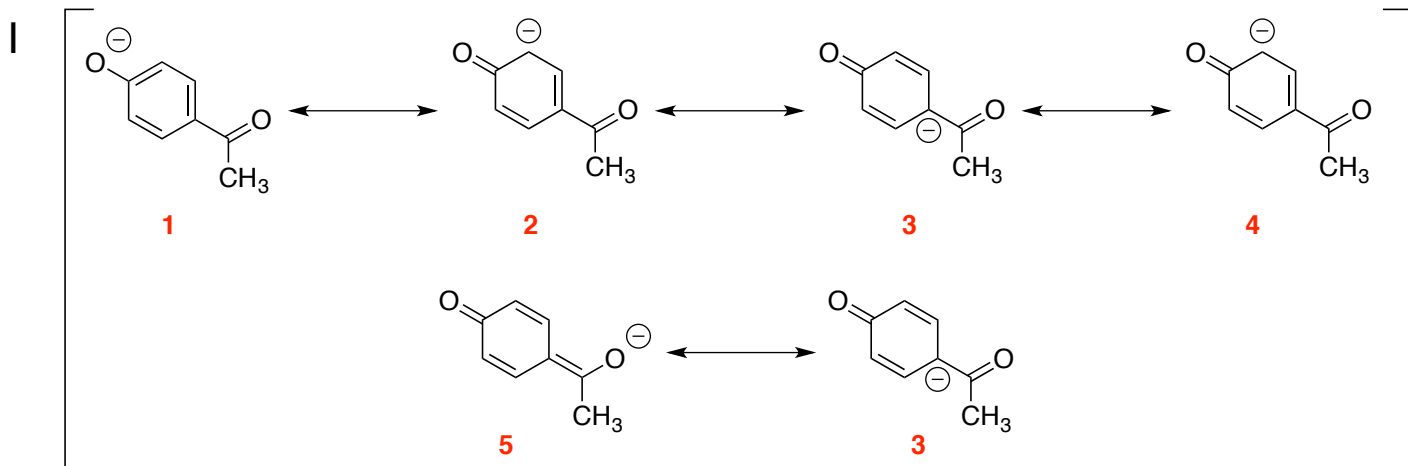


Cation Resonance Structures





Due to the positive charge on the electronegative atom (N), the lower resonance hybrid is of higher energy



The charge is spread out over 6 carbons!