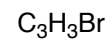
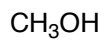
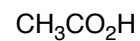
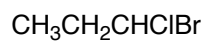


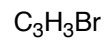
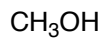
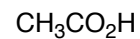
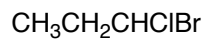
Supplemental Instruction Worksheet-Week 1

1. Draw Lewis structures for the following.

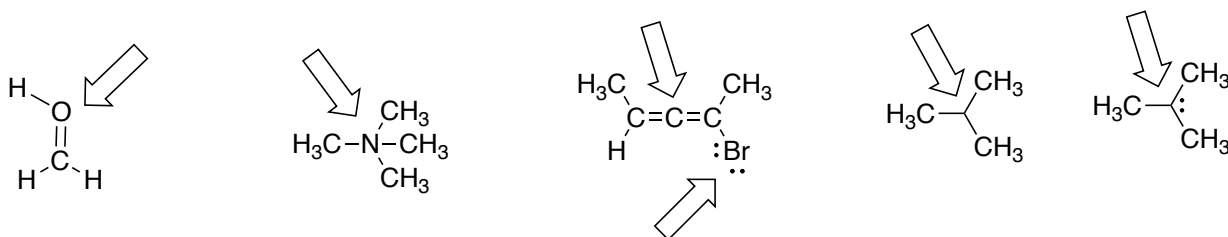
Come see me or Logan for Answers



2. Now draw bond-line structures for each of the compounds above

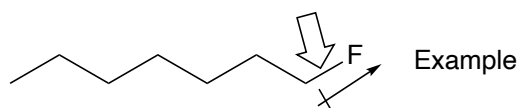


3. Provide the formal charge on each of the indicated atoms below. Note, in organic chemistry the choice is +1, -1 or zero for most organic structures and atoms used in organic chemistry.



Come see me or Logan for Answers

4. Indicate the direction of the dipole using \rightarrow for each of the indicated bonds.



Come see me or Logan for Answers
Also, here look at hybridization and sigma/pi bonds

